CHANGE IN CHARGING STATE OF INSULATORS IRRADIATED BY CHARGED PARTICLES OF LOW AND MEDIUM ENERGY

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Introduction

Study of charging of dielectric materials under electron and ion irradiation is not only of scientific, but also of great practical importance, for example in methods of surface analysis using ion and electron beams, scanning electron microscopy and focused ion beam microscopy, in lithography and space technology.

Electron irradiation

It was found experimentally [1, 2] that the charged surface potential of some dielectric samples (for example, crystalline sapphire) begins to grow not immediately after electron irradiation switching on, but only after a certain critical irradiation fluence $F_{\rm cr}$.

Ion irradiation

Previous experiments [3] showed that under positive Ar^+ ion bombardment the surface of insulators is charged to a very high positive potential. It was also observed a significant increase in the yield of positive ions emitted from the surface. The purpose of recent study was to investigate the reasons for such an ion emission enhancement under medium energy ion irradiation of insulators charged to high potentials.

Materials and Methods

1 - ion gun

Experimental setup

In present research the kinetics of charged surface potential V_S and intensity of cathodoluminescence signal I_{CL} was measured simultaneously.

Materials and Methods

Samples (100) SiO_2 and (0001) Al_2O_3 single crystals

Experimental setup



- 1 objective lens
- 2 electron beam
- 3 dielectric sample
- 4 conductive substrate
- 5 grounded metal grid
- 6 hemispherical
- electron collector
- 7 electrostatic toroidal spectrometer
 10 cathodoluminescence signal
 detector
- D_1 , D_4 , D_2 , D_3 signal detectors

Parameters of electron beam

• electron beam energies $E_0 = 5$ and $E_0 = 15$ keV



- Parameters of ion beam
 ion beam energy E₀= 6 keV
- ion beam current $I_0 = 500 \text{ pA}$

2 - dielectric target 3 - metal substrate 4 - insulator 5 - shielding chamber for samples 6 - toroidal sector energy analyzer 7 - energy analyzer spectrometer 8 - hemispherical ion collector 9 - microchannel plate detector (MCP) D₁, D₂, D₃ - collimating diaphragms.

Experimental results

Time evolution of charging characteristics of pristine SiO₂ crystal under Ar⁺ ion irradiation



 I_0 – incident ion current I_{σ} -emission current I_{L+D} - displacement current V_S – surface potential

- electron beam current density $j_0 = 10^{-5} \text{ A/cm}^2$
- irradiated area $S = 100 \times 100 \ \mu m^2$

Experimental results

Cathodoluminescence intensity I_{CL} and surface potential $-V_S$ of Al_2O_3 crystal as a function of electron irradiation time t

pristine Al₂O₃ crystal



Conclusions

• It has been found a non-monotonic dependence of cathodoluminescence intensity from pristine Al_2O_3 crystal on the electron irradiation time.

• For the pristine sapphire crystal, there is a critical fluence of electron irradiation at which the charging potential starts to grow. The onset of charging coincides in time with the onset of the decay of the cathodoluminescence signal.

• The critical fluence at which the charging of the sapphire crystal begins is inversely proportional to the electron irradiation energy due to the inverse proportionality of the inelastic interaction cross section to electron energy.

• For sapphire pre-irradiated by electrons with high fluence of 2×10^{20} cm⁻², charging begins immediately after electron irradiation switching on and a monotonic dependence of cathodoluminescence signal is observed.

• These experimental results can be explained by a generation of radiationinduced defects which are the luminescence centers in crystal. At high electron fluence the radiation defect concentration reaches the great magnitude and defects start to form aggregates which are the deeper traps for electrons. It leads to the decay of the cathodoluminescence signal and the onset of charging.

Al₂O₃ crystal pre-irradiated by electrons energy 1 keV, fluence 2 ×10²⁰ cm⁻²



• The observed enhancement of ion-ion emission from insulators charged to high potentials can be explained by a decrease in probability of primary ion neutralization and a reduction of the chemical bond energy due to appearance of a hole at the O atom in the lattice.

References

[1] E.I. Rau, A.A. Tatarintsev, E.Yu. Zykova. *NIMB* 460 141, 2019.
[2] E.I.Rau, A.A. Tatarintsev, E.Yu.Zykova, et al., *Phys. Solid State* 59 1526, 2017.
[3] E. Rau, A. Tatarintsev, E. Zykova, et al., *Vacuum* 177 109373, 2020.